

Endothermic vs Exothermic Worksheet

1. Define exothermic and endothermic.

exothermic:

endothermic:

2. Ice cube melting on a counter top:

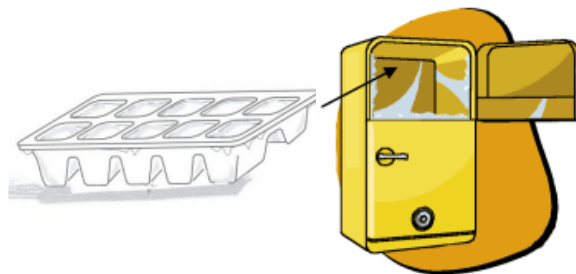
- a. Draw arrows showing which direction heat is being transferred (in or out of) the ice cube.
- b. For the ice cube, is this process endothermic or exothermic? Explain.



- c. For the surroundings, is the process endothermic or exothermic? Explain.

3. Water freezing in an ice cube tray in the freezer:

- a. Draw arrows showing which direction heat is being transferred when the ice cube tray is placed in the freezer.
- b. For the water, is the process endothermic or exothermic? Explain.



- c. In order for things to cool in a refrigerator, heat must be transferred away from the food. Where does this heat end up going?

4. Boiling water on the stove:

- a. Draw arrows showing which direction heat is being transferred in the image to the right.
- b. Water (l) to water (g): is this process endothermic or exothermic? Explain.



- c. When is it colder outside? (a) while its raining or (b) after it rains? Explain.

5. State whether the six phase changes are endothermic or exothermic.

Melting: _____

Freezing: _____

Boiling: _____

Condensing: _____

Subliming: _____

Depositing: _____

ON THE BACK:

6. Draw the potential energy diagrams for an endothermic and an exothermic system.